A Novel Synthesis of 3,4-Dihydro-(2H)-1,2,4-Benzothiadiaizine-1,1-Dioxides Promoted by Samarium Diiodide

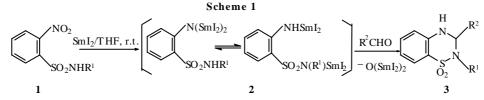
Wei Hui ZHONG, Yong Min ZHANG*

Department of Chemistry, Zhejiang University (Campus Xixi), Hangzhou 310028

Abstract: 3,4-Dihydro-(2H)-1,2,4-benzothiadizine-1,1-dioixdes were prepared in good yields *via* reductive cyclization of o-nitrobenzenesulfonamides with aldehydes promoted by SmI_2 under mild and neutral condition.

Keywords: Samarium diiodide, 2H-1,2,4-benzothiadizine-1,1-dioxide, reductive cyclization.

Applications of samarium diiodide as a mild, neutral, selective and versatile single-electron transfer reducing and coupling reagent in organic synthesis have significantly grown in the last decade¹. Previous work on some deoxygenation and some reductive cleavage of S-S, Se-Se, Te-Te bonds with SmI₂ was reported². It is well known that nitro compounds can be reduced easily by SmI₂. However, they are just reduced to the corresponding amines³, little attention has been concerned on the intermediate derived from a nitro group by SmI₂ treatment, which may be induce some reactions difficult to accomplish by other existing methodologies⁴. Herein, we wish to report the preparation of 3,4-dihydro-(2H)-1,2,4-benzothiadizine-1,1-dioixides in one pot *via* reductive cyclization of *o*-nitrobenzenesulfonamides with aldehydes promoted by SmI₂ under mild and neutral condition.



The results are summarized in **Table 1**. When *o*-nitrobenzenesulfonamides **1** were treated with SmI_2 at room temperature, the reduction of nitro group resulted in the formation of the trivalent samarium species. According to the relative literature⁴, we considered these species may be samarium amides **2**, which are "living" anions and reacted smoothly with aldehydes to afford 3-substituted 3,4-dihydro-(2H)-1,2,4-benzothiadizine-1,1-dioxides **3** in moderate to high yields.

2H-1.2.4-benzothiadizine-1,1-dioxide derivatives have attracted strong interest due to their biological properties such as their diuretic activity^{5a}, antihypertensive activity^{5b}. Several methods had been introduced to prepare this kind of compounds⁵. Here we

Wei Hui ZHONG et al.

provided novel route to synthesize а 3,4-dihydro-(2H)-1,2,4-benzothiadizine-1,1-dioxides, the advantage of which are readily available starting materials, straightforward and simple synthetic procedures, mild reaction conditions and moderate to high yield.

Entry	\mathbb{R}^1	R^2	Reaction condition		Yield(%) ^b
			T∕°C	t / h	
3a	Н	n-Pr	r.t.	2	90
3b	Н	n-Bu	r.t.	2	85
3c	Н	n-C ₆ H ₁₃	r.t,	3	83
3d	Н	$C_6H_5CH_2$	r.t.	3	76
3e	Н	C_6H_5	r.t	3	80
3f	Н	p-MeOC ₆ H ₄	r.t	3	81
3g	Н	m-BrC ₆ H ₄	r.t	3	83
3h	Н	p-NO ₂ C ₆ H ₄	r.t	3	72
3i	Me	n-Pr	40	3	65
3j	Me	C ₆ H ₅	40	3	60

Table 1 Reaction of o-nitrobenzenesulfonamides with aldehydes promoted by SmI₂^a

^a 1 equiv. nitro compounds, 1.2 equiv.aldehydes and 6 equiv. SmI₂ were used; ^b isolated yields based on nitro compounds; all the products were characterized by ¹H NMR and IR spectra.

General procedure: A solution of o-nitrobezenesulfomamides 1 (1 mmol) in anhydrous THF (2 ml) was added dropwise to a solution of SmI_2 (6 mmol) in THF (20 ml) at room temperature under dry nitrogen atmosphere. The mixture was stirred for 0.5 hrs and became yellow gradually. Then aldehydes (1.2 mmol) in THF (2 ml) were added. After stirred for a given time (Table 1), the reaction was quenched with dilute hydrochloric acid (0.1mol/L, 3 ml). The crude product was isolated with usual ways and purified by preparative TLC on silica using with ethyl acetate and cyclohexane (1: 2) as eluent.

Acknowledgment

We thank the National Natural Science Foundation of China (Project No.29872010) and the Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences for financial support.

References

- For reviews see: a) G. A. Molander, C. R. Harris, Chem. Rev., 1996, 96, 307. b) T. Imamoto, 1. Lanthanides in organic Syntheis, Academic Press: London, 1994, Chapter 4. c) G. A. Molander, Chem, Rev., 1992, 92, 29.
- a) S. Fukuzawa, Y. Niimoto, T. Fujinami, S. Sakai, *Heteroatom Chem.*, **1991**, *1*, 490. b) Y. M.
 Zhang, R. H. Lin, *Synth. Commun.*, **1987**, *17*, 329. c) Y. M. Zhang, Y. P. Yu, R. H. Lin, *Synth. Commun.*, **1993**, *23*, 189. d) H. J. Jiang, Y. M. Zhang, *Chinese Chem, Lett.*, **1999**, *10*, 7.
 a) P. Girard, J. L. Namy, H. B. Kagan, *J. Am. Chem. Soc.*, **1980**, *102*, 2693. b) J. Souppe, L. 2.
- 3. Danon, J. L. Namy, H. B. Kagan, J. Organometal. Chem., 1983, 250, 227.
- 4.
- L. H. Zhou, Y. M. Zhang, J. Chem. Soc. Perkin Trans. I, **1998**, 2899. a) C. T. Holderge, R. B. Babel, L. C. Cheney, J. Am. Chem. Soc., **1959**, 81, 4807. b) L. H. 5. Werner, A. Halamandaris, S. Ricca, Jr., L. Dorfan, G deStevens, J. Am. Chem. Soc., 1960, 82,

1161. c) J. H. Freeman, E. C. Wagner, *J. Org. Chem.*, **1951**, *16*, 815. d) F. C. Novello, S. C. Bell, F. L. A. Abrams, C. Ziegler, *J. Org. Chem.*, **1960**, *25*, 970.

Received 3 January 2000